

Scale-Up Radiolysis Testing of Aluminum-clad SNF and Surrogates in a Mini-Canister Environment (part of Task 6)

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Radiolytic hydrogen generation

- Aluminum oxyhydroxides form on aluminum exposed to water
 - On fuel cladding during pre-treatments, in-reactor, and/or storage
 - Types
 - Bayerite or gibbsite, $Al(OH)_3$ or $Al_2O_3 \cdot 3H_2O$, forms at low temperature (<80°C)
 - Boehmite, AlOOH or $Al_2O_3 \cdot H_2O$, forms at higher temperature
- Hydrogen (H₂) gas can be generated in dry storage canisters for aluminum-clad spent nuclear fuel (SNF) due to radiolysis of
 - Residual liquid water/water vapor
 - Physisorbed water
 - Aluminum oxyhydroxides (chemisorbed water)
- Potential for pressurization or flammable gas mixture
- Expected yield and rate of production needed for performance prediction and the safety basis

Mini-canister radiolysis testing

- Measure radiolytic gas generation in canister-analogous environment
 - Stainless-steel vessel
 - Helium cover gas (25 psia)
 - Lab-grown surrogates or reactor-exposed aluminum samples bearing Al oxyhydroxides
- On-line monitoring (multiple measurements over time for same test)
- Test data used to
 - Evaluate impact of drying
 - Benchmark models

Mini-canister

- Commercial small canisters
- Gas sampling ports on top connect to instrumentation



Loaded and sealed mini-canister with reactor-exposed sample

Tested in Co-60 (gamma) irradiator

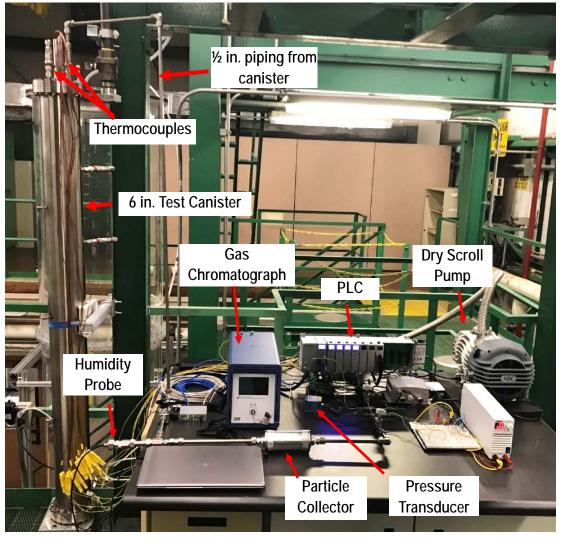
- Stand-in for spent fuel environment
- Dose rate for samples: 37.8 krad/hr (gamma)
- Ambient irradiator temperature: ~23.5°C
- Gas sampled at intervals during test
- Gas monitoring system
 - Two-column gas chromatograph: Ar and He carrier gases
 - 20-m Ar column for greater separation of H₂ from He cover gas
 - H₂ sensitivity down to 100 ppm
 - N₂ and O₂ sensitivity down to 20 ppm

On-line monitoring instrumentation

 System developed for future monitoring of an actual ASNF dry storage "demo" canister

Instrumentation

- Valves, tubing, and vacuum pump to withdraw sample
- Particle collector to capture particulates
- Ion chamber to measure radioactivity in exiting gas
- Gas characterization
 - Humidity sensor
 - Gas chromatograph
 - Pressure transducer
- He tank for optional topping off He
- HEPA filter before exhausting sample gas
- Data acquisition system

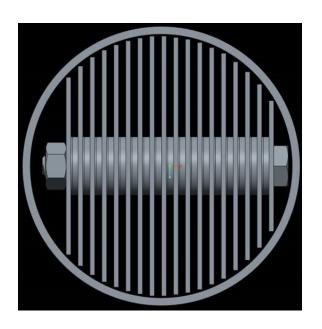


Instrumentation connected to 6-in. diameter test canister

Samples

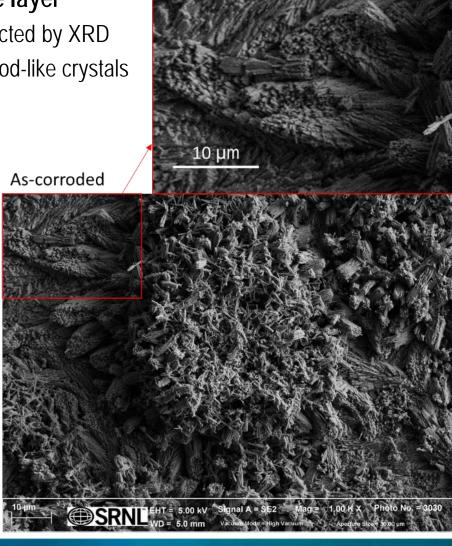
Surrogate coupon assemblies

- Parallel plates held together by bolt
- Spacers to maintain gap
- Maximizes sample area in mini-canister
- Corroded in-lab
 - Room-temperature water
 - -30+ days



Oxyhydroxide layer

- Bayerite detected by XRD
- SEM shows rod-like crystals
- ~9 μm thick



Mini-canister loading

- Assembly loaded and sealed in canister
- Residual air elimination
 - Held under vacuum 12 h
 - Four He purges (alternating helium fill and vacuum)
- Helium backfill to 25 psia
- Two tests with identical assemblies
 - "As-corroded" (no further drying)
 - "As-dried" (additional heated drying step)



As-dried assembly

Thermal drying

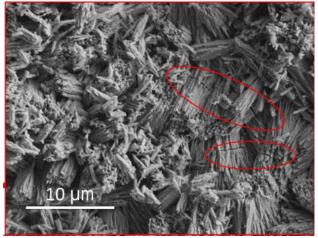
- Sampling line left open on sealed canister
- Canister heated to 220°C in air for 4 h
- Vacuum for 1 h at 220°C
- Residual air elimination and He backfilling procedure

Impact of drying on oxyhydroxide

- Reduced bayerite peaks and increased boehmite peaks in XRD
- Caused minor cracking



Visible water release through sampling line during heating



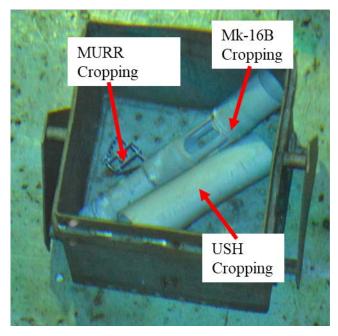
Small cracks in oxide after drying

- Retrieved from L-Basin wet storage at Savannah River Site
 - Aluminum pieces irradiated in-reactor
 - Do not contain fuel



• From Missouri University Research Reactor (MURR)

- Al-6061
- Operating conditions: >60°C, 113 days
- Wet storage <18 years at ~30°C

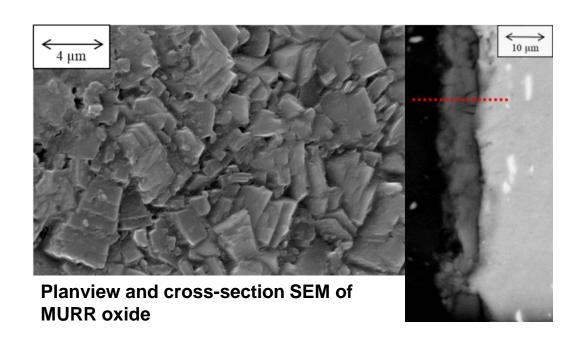


Sampled materials collected in L-Basin

MURR in mini-canister

MURR oxyhydroxide layer

- Bayerite and boehmite detected by XRD
- Relatively uniform blocky oxide
- $\sim 10 \ \mu \text{m} \text{ thick}$



No drying except drip-drying in air

- Provides bounding H₂ rate for this sample
- Helium purges to remove residual air

Small sample compared to surrogates

- Less H₂ generation expected
- Larger gas volume → H₂ more diluted

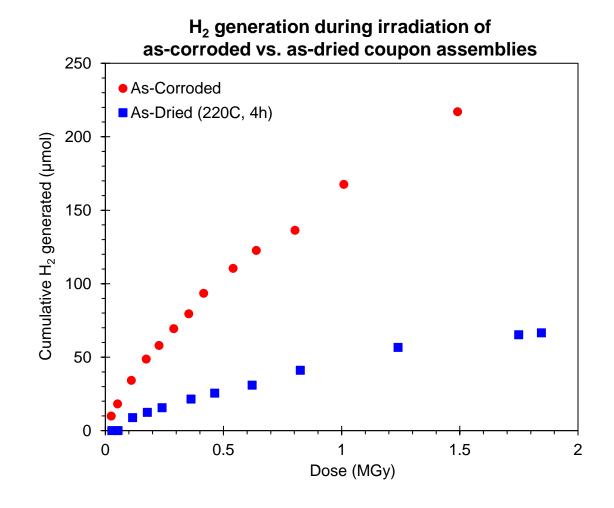


MURR in mini-canister

Results

Results: Surrogate coupon assemblies

- H₂ generation rate much lower for as-dried than ascorroded surrogate
- Large difference in overall H₂ yield
 - As-dried assembly appears to be leveling off by ~2 MGy dose
 - As-corroded assembly yield ~3x larger at ~1.5 MGy and still climbing
- Heated drying of ASNF canisters to 220°C has potential to greatly reduce H₂ generation

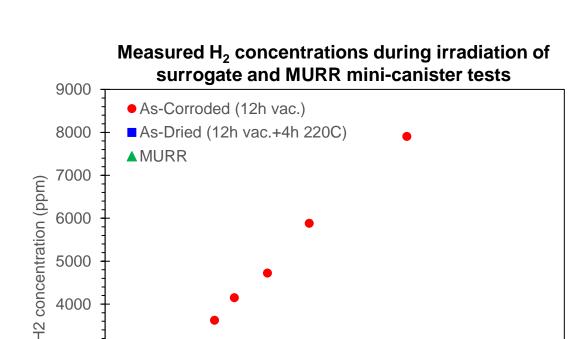


4000

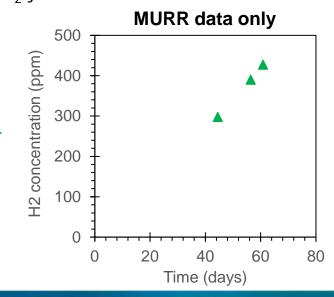
3000

2000

1000



- Small but increasing H₂ concentration observed
 - Smaller than for surrogate assemblies at same time/dose
 - Smaller sample requires longer to accumulate H₂
- Preliminary calculations suggest H₂ generation per unit surface area greater than as-corroded assembly
 - Consistent with lack of thermal or vacuum drying for MURR
 - Recent ampoule tests showed 12-h vacuum significantly reduced H₂ yield



100

Time (days)

50

200

250

150

Conclusions and path forward

- As-corroded and as-dried surrogate tests demonstrate that elevated-temperature drying can reduce radiolytic hydrogen generation
 - As-dried sample had
 - Lower H₂ generation rate from beginning of test
 - Apparent plateau or transition to much lower generation rate at fraction of the H₂ yield of the as-corroded sample
 - This data will be used in combination with drying experiment results to select drying parameters for dry storage applications

Modeling

- Data on the H₂ generation rates and plateau will be used to compare to model predictions
- H₂ generation rates for the as-dried condition can be incorporated into the model to enable estimation of canister conditions following an appropriate drying condition.
- MURR test to continue to obtain H₂ yield curve for reactor-exposed material
 - Provides validation for H₂ generation rates obtained from surrogates

- Task Technical and Quality Assurance Plan for Development of the Instrumented Lid for Dry Storage of Aluminum Spent Nuclear Fuel, SRNL-RP-2019-00225, March 2019, D. Herman and J. McNamara.
- Instrumented Lid System Final Design, SRNL-L2240-2019-00002, May 2019, J. McNamara and B. Randall.
- Instrumented Lid Detailed Design, SRNL-L2240-2019-00003, June 2019, J. McNamara and A. McNight.
- Instrumented Lid Assembly of System Components, SRNL-L2240-2019-00024, August 2019, J. McNamara, D. Pak, A. McNight.
- Instrumented Lid System Functionality Testing, SRNL-L2240-2019-00004, September 2019, J. McNamara and B. Randall.
- Instrumented Lid Fabrication of Basket, SRNL-L2240-2019-00006, December 2019, J. McNamara, D. Pak, A. McNight.
- Status Update on the Instrumented Lid Project, SRNL-L2240-2019-00007, December 2019, J. McNamara, B. Randall, D. Pak, A. Duncan, A. McNight, H. Sessions, A. Baldwin, D. Hunter
- Instrumented Lid Full Instrumentation System Testing, SRNL-L2240-2019-00005, Rev. 1, February 2020, J. McNamara, B. Randall, D. Pak, A. Duncan, A. McNight, H. Sessions, A. Baldwin, D. Hunter.

Reports: Mini-canister radiolysis experiments

 Aluminum Spent Nuclear Fuel 2020 Large Sample Radiolysis and Measurement Test Plan, SRNL-RP-2020-00219, April 2020, J. McNamara and C. Verst.

- Instrumented Lid Fabricate Hydrated Oxide Specimens for Testing, SRNL-L6000-2020-00034, August 2020, J. McNamara and C. Verst.
- Instrumented Lid Initiation Irradiation and Measurement of As-is Hydrated Oxide Specimens (Large Coupons),
 SRNL-L6000-2020-00034, August 2020, J. McNamara and C. Verst.
- Initiation Irradiation and Measurement of Dried Hydrated Oxide Specimens (Large Coupons), SRNL-L6000-2020-00046, September 2020, C. Verst, B. Randall, J. McNamara.
- Interim Irradiation and Measurement of As-dried vs As-corroded Hydrated Oxide Specimens (Large Coupons), SRNL-L6000-2021-00006, March 2021, C. Verst, A. d'Entremont, B. Randall, J. McNamara.

Thank you!